

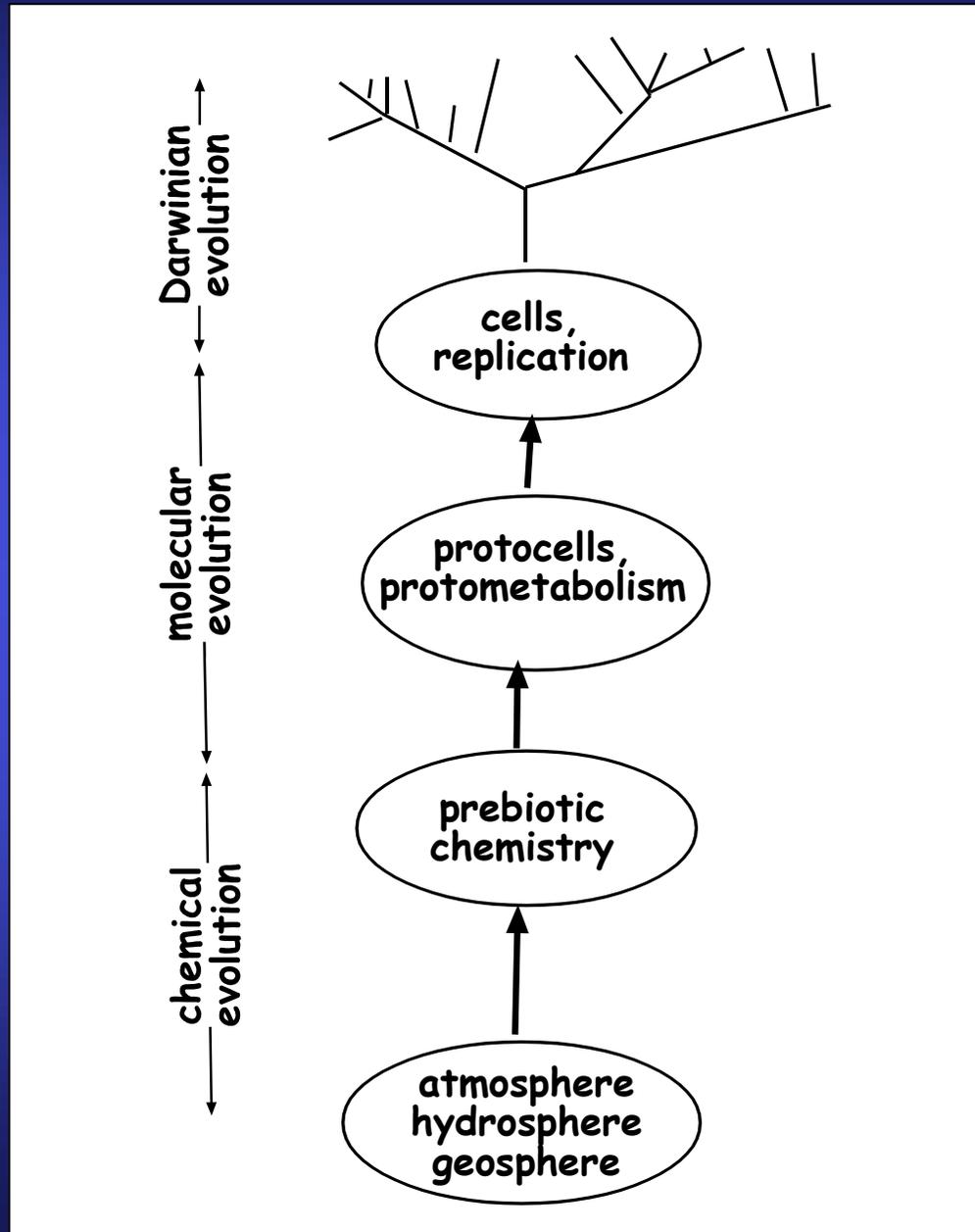
Hydrothermal systems: the site of the origin and early evolution of life?

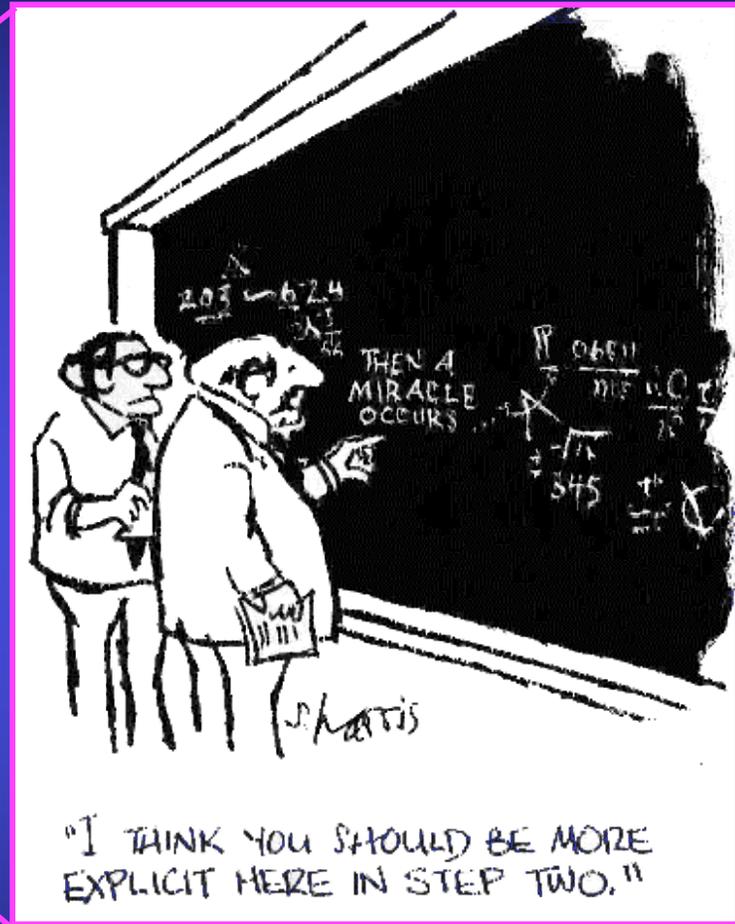
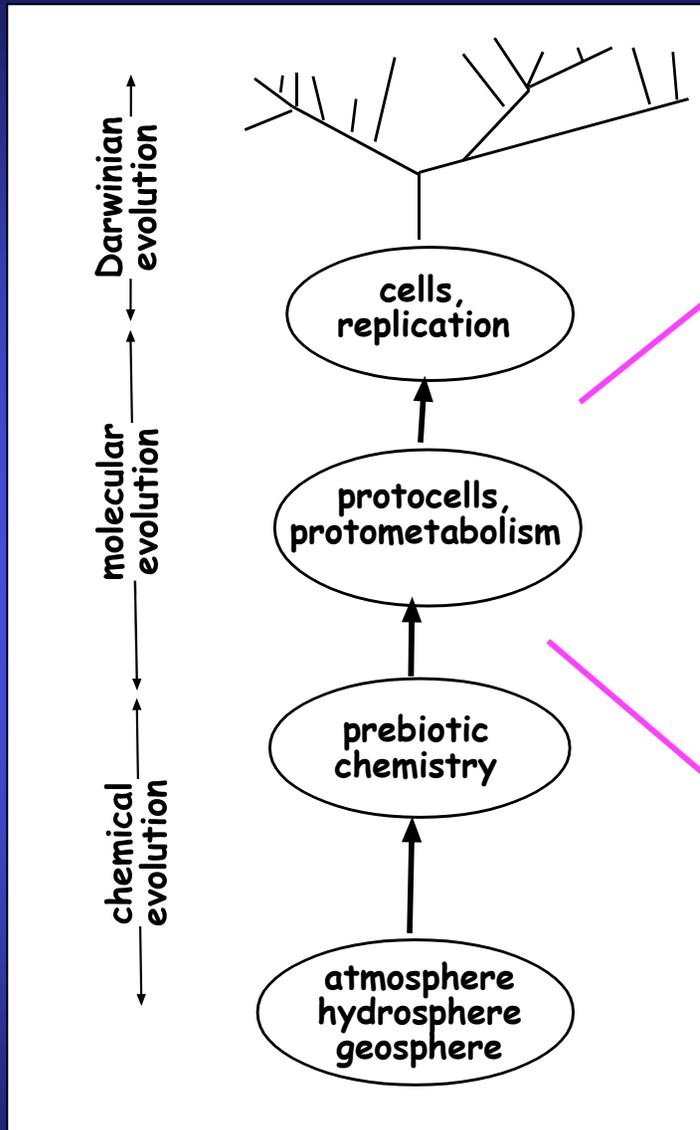
Tom McCollom

The Center for
Astrobiology
at the University of Colorado

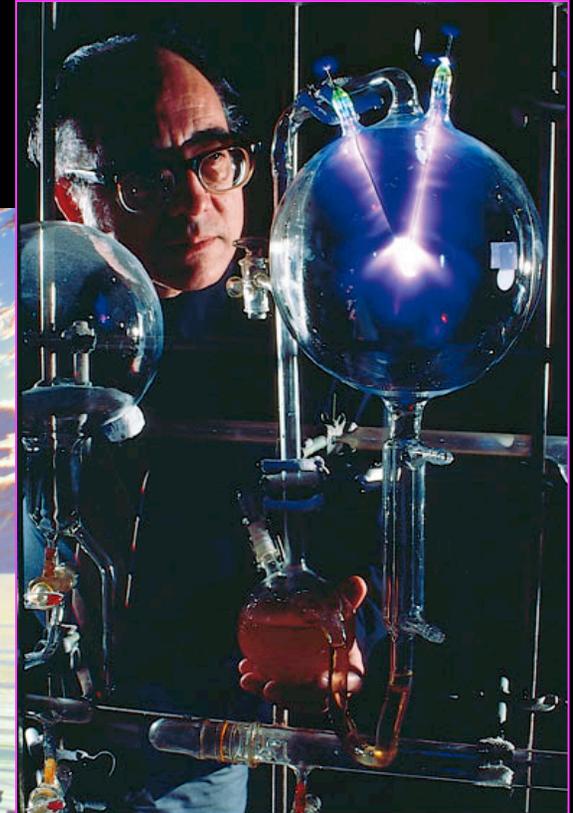
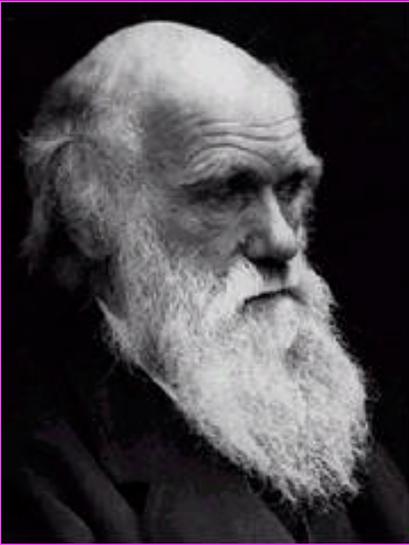


Origin of life: The grand scheme





The origin of life: A primordial soup in a "warm little pond"?



**The origin of life:
... or maybe in a hydrothermal system?**

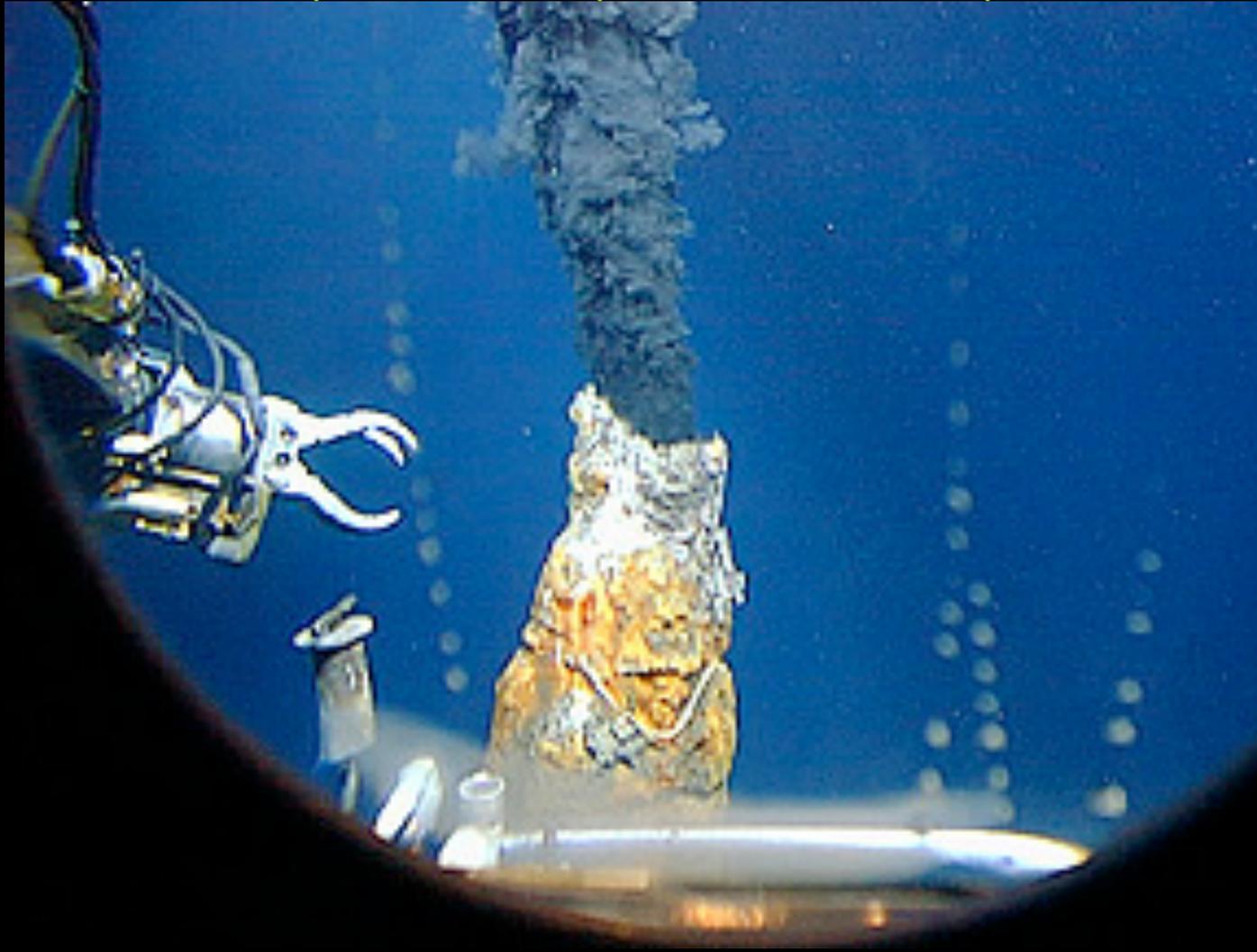


Photo by Patrick Hickey

A basic dichotomy in theories of the origin of life:

Heterotrophic origin – first organisms consumed organic compounds present in their milieu (the “primordial soup”)

Autotrophic origin – first organisms derived energy from chemical reactions and produced their own biomass

Many features of hydrothermal systems would have been conducive to the origin & early evolution of life:

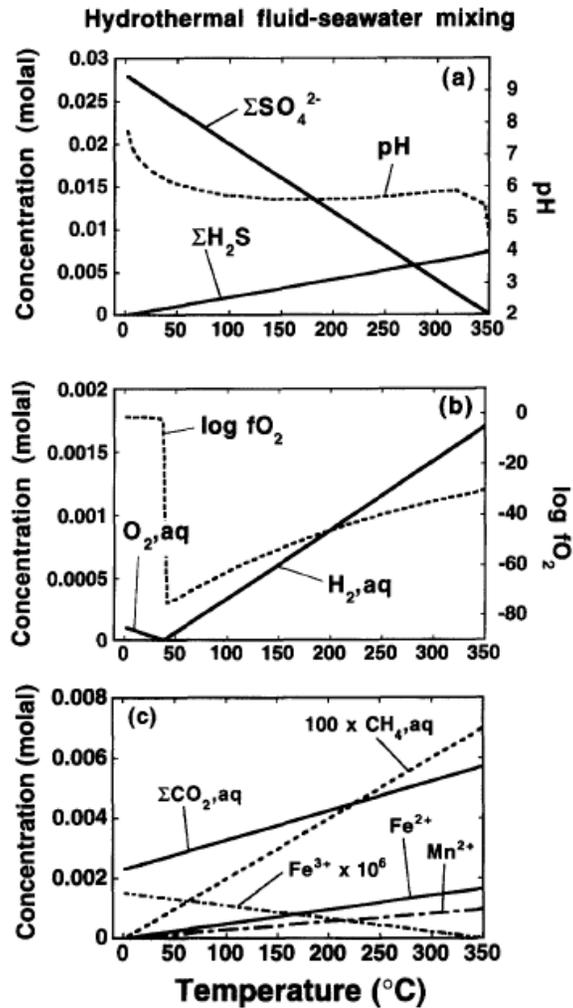
- ◆ Chemical & temperature gradients favorable for chemical evolution
- ◆ Conditions favorable for abiotic synthesis of prebiotic organic compounds
- ◆ Mineral surfaces promote organic synthesis & chemical evolution

Hydrothermal systems create steep chemical and physical gradients and chemical disequilibria

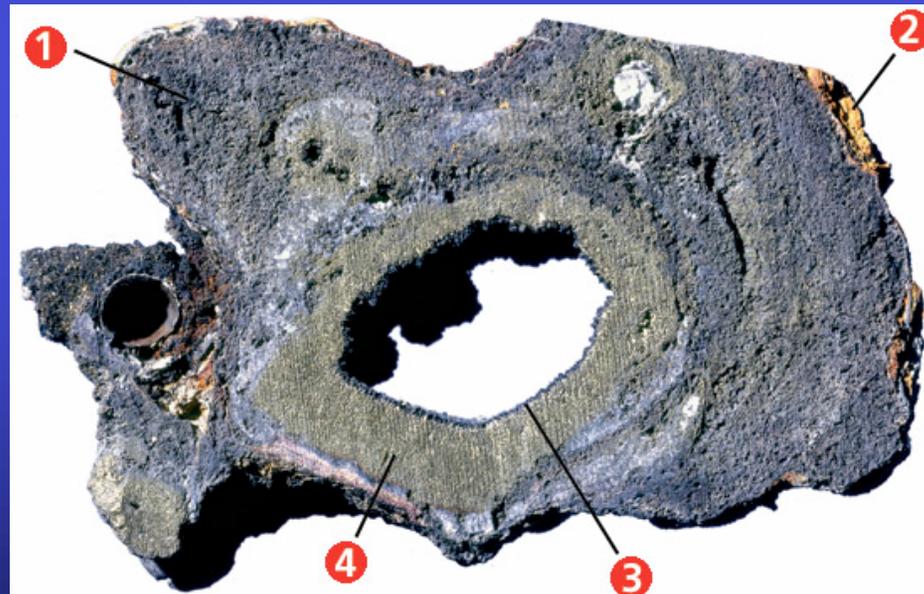
- ◆ favorable environments for chemical reactions (particularly oxidation-reduction reactions)
- ◆ provides sources of chemical energy for metabolism



Example: mixing of hydrothermal fluids with seawater at a deep-sea hydrothermal vent chimney

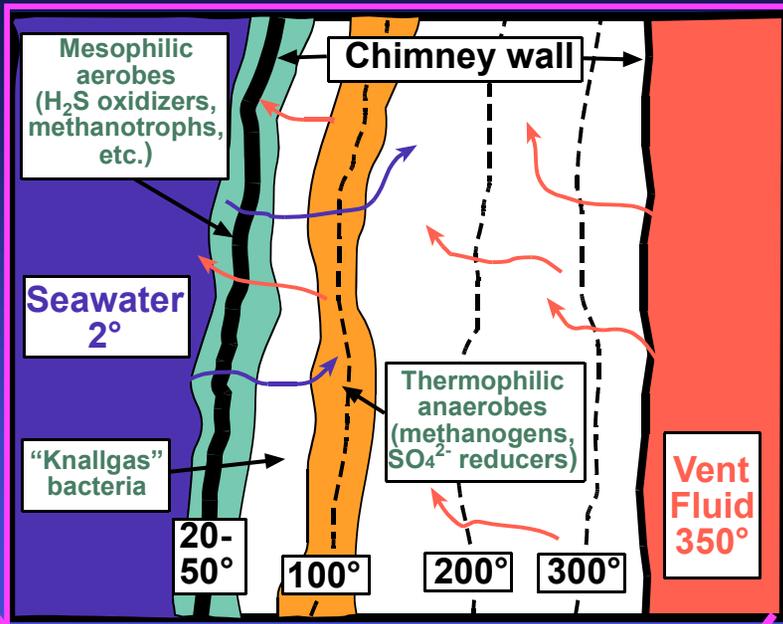


Modeled fluid composition for mixing of hydrothermal fluid & seawater across chimney

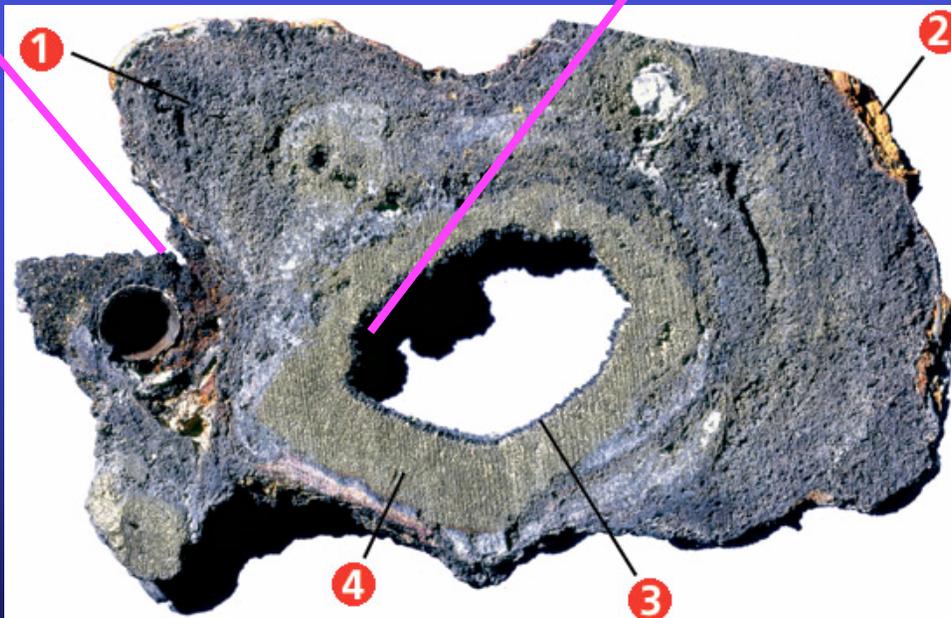


Cross section of deep-sea hydrothermal chimney

(Cross-section image courtesy Meg Tivey, Woods Hole Oceanographic Inst.)



Chemical & thermal gradients provide a variety of chemical energy sources leading to multiple microbial niches in chimney walls



Chemically reducing conditions in hydrothermal environments conducive to conversion of inorganic compounds (or simple organics) to prebiotic organic compounds:



Lipids, amino acids, sugars, nucleotides(?), etc.

E.g., formation of simple sugars and amino acids from formaldehyde

(Weber, *Origins of Life & evolution of the Biosphere*, 2001)

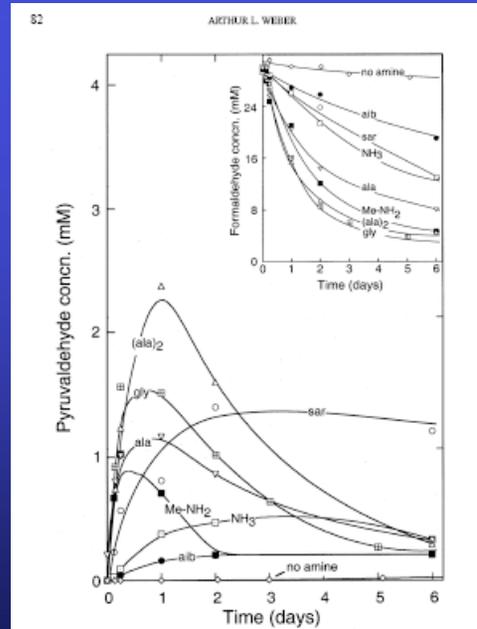
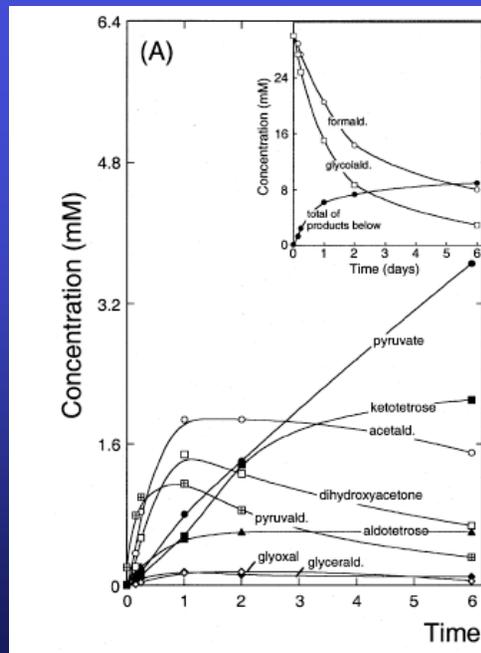
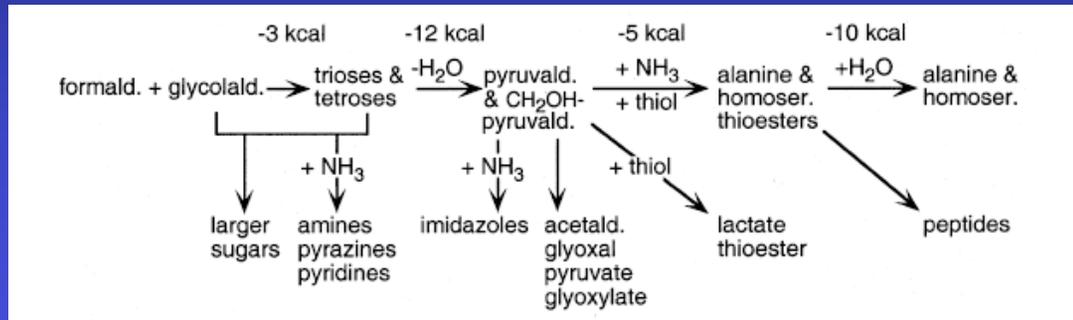
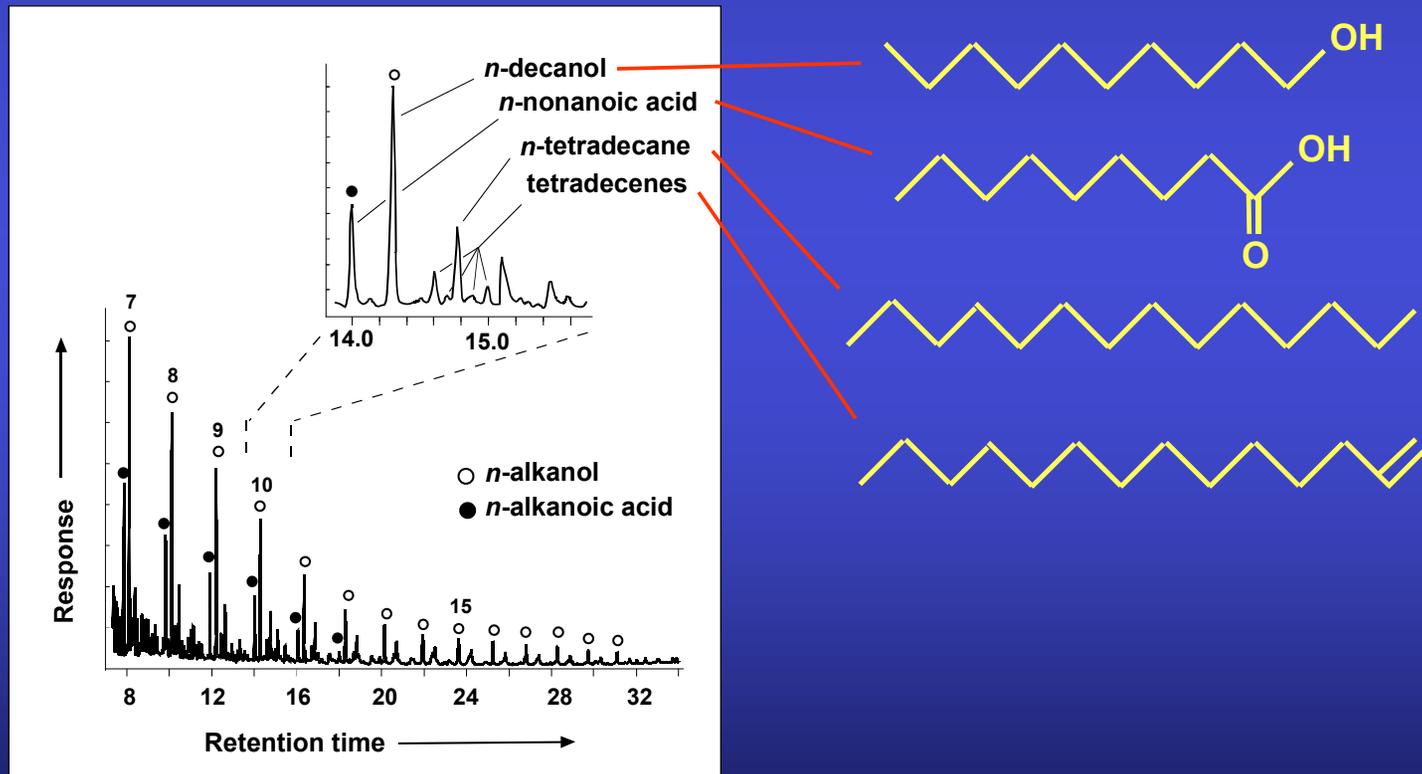


Figure 3 Time course of the amine-catalyzed formation of pyruvaldehyde from 30 mM formaldehyde and 30 mM glycolaldehyde in 200 mM sodium acetate (pH 5.5) in the dark under anaerobic conditions at 50°C. The reactions used 15 mM α -aminobutyric acid (alb), sarcosine (sar), serine (Ser), alanine (ala), methylamine (Me-NH₂), diethylamine (diEt), and glycine (gly). The plot inserted in the upper right of the figure shows the formaldehyde and glycolaldehyde concentrations during the reaction.

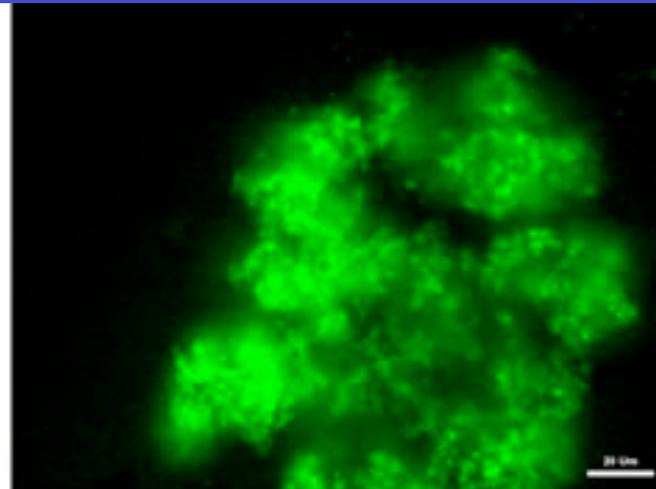
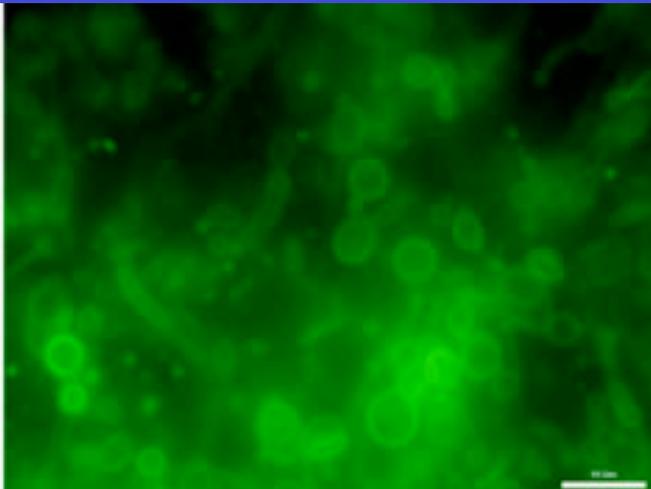
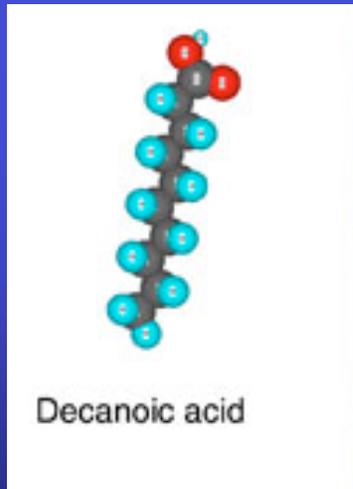
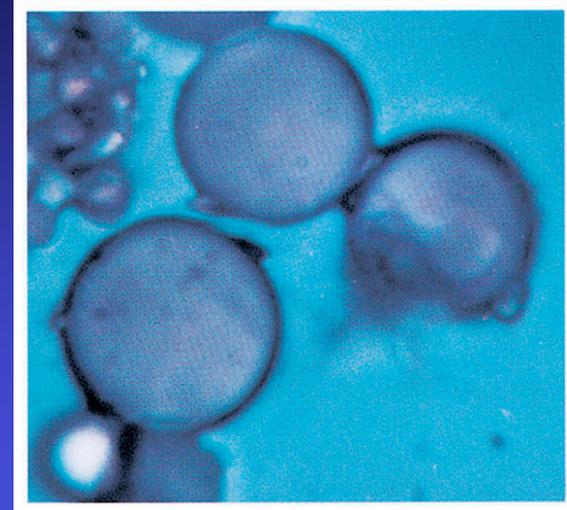
E.g., lipid synthesis under hydrothermal conditions

Fischer-Tropsch reaction products
(Formic acid in H₂O, 72 hours @ 175°C)



(McCollom et al., 1999)

Lipids spontaneously form micelles (lipid-bound cells) during cooling of hydrothermal solutions:
The first cellular membrane??



(Monnard et al. *Astrobiology*, 2004;
http://www.chemistry.ucsc.edu/deamer_d.html)

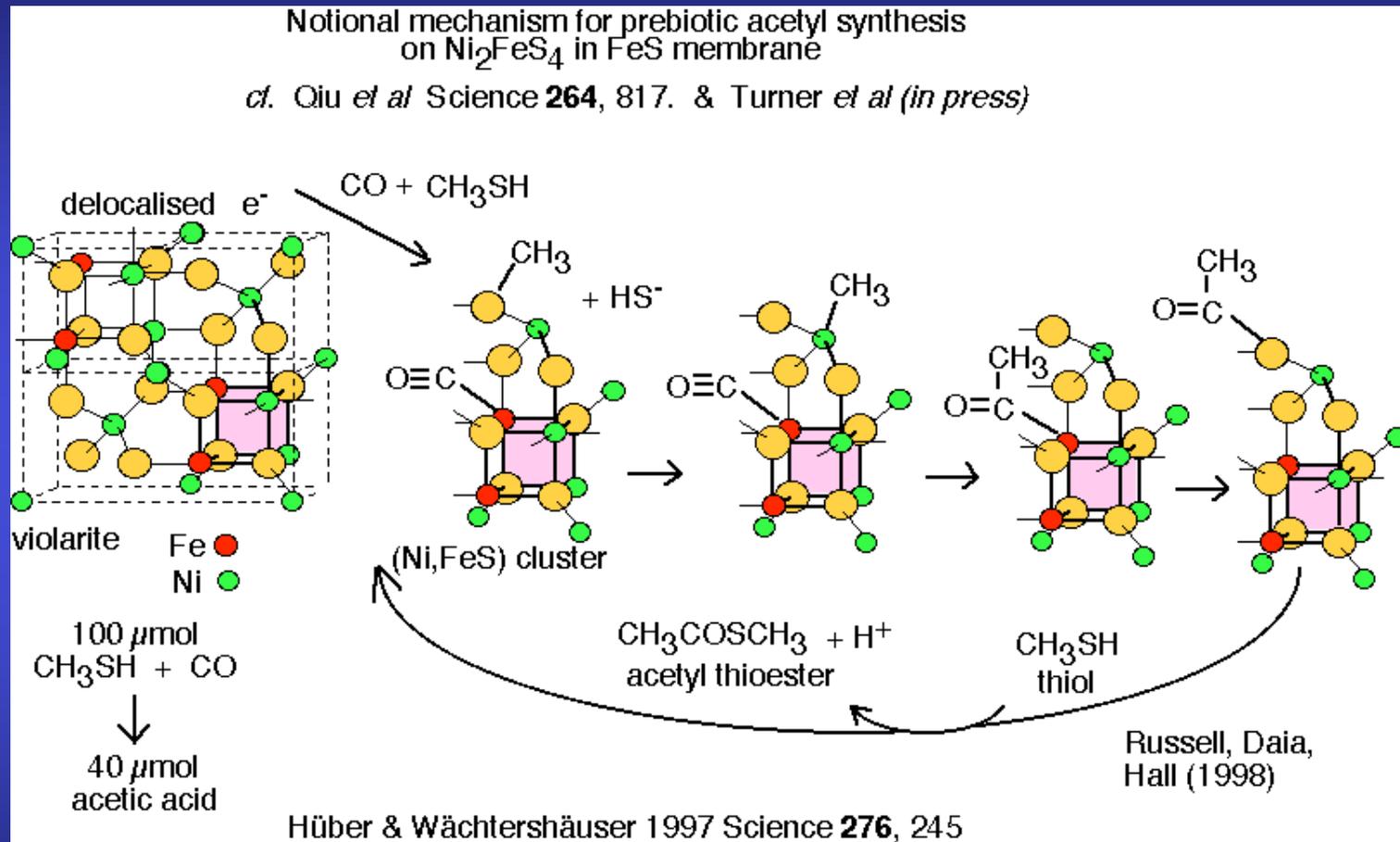
Minerals precipitated from hydrothermal solutions (sulfides, clays) promote chemical evolution through catalysis & templating



ALV 2462
RO2
1 cm

(image courtesy Meg Tivey, Woods Hole Oceanographic Institution)

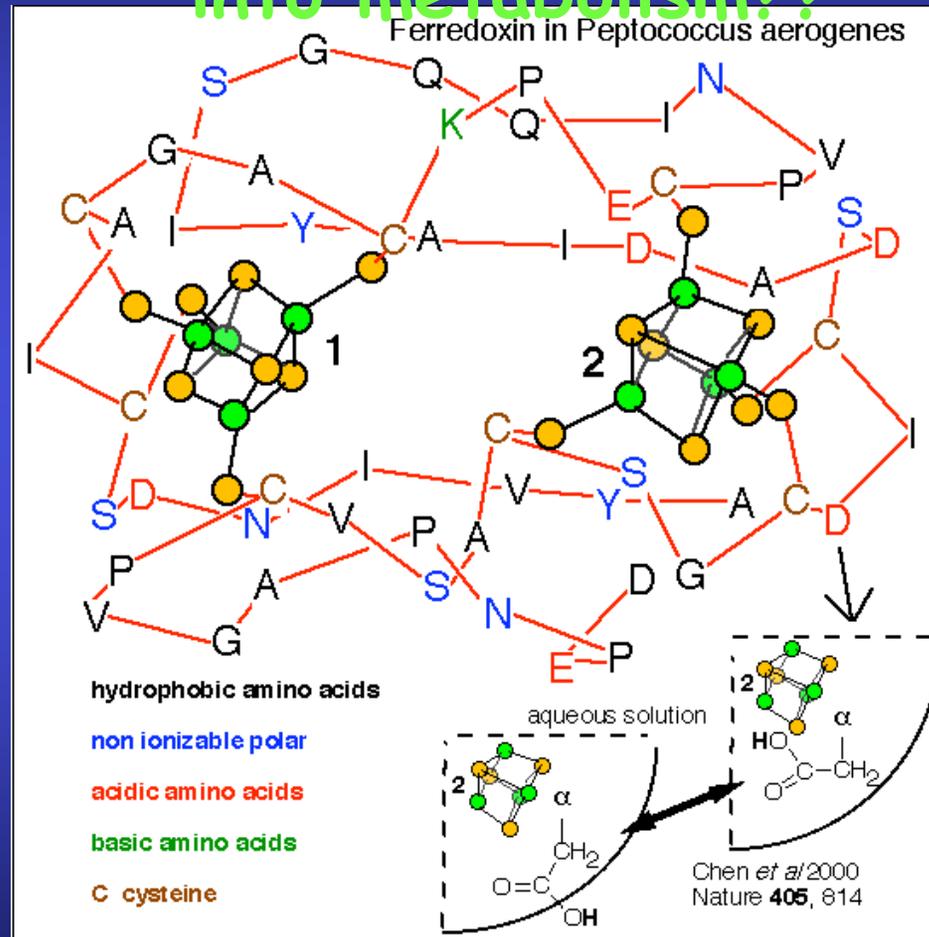
E.g., sulfide mineral catalysis of acetic acid:



(image courtesy Mike Russell;

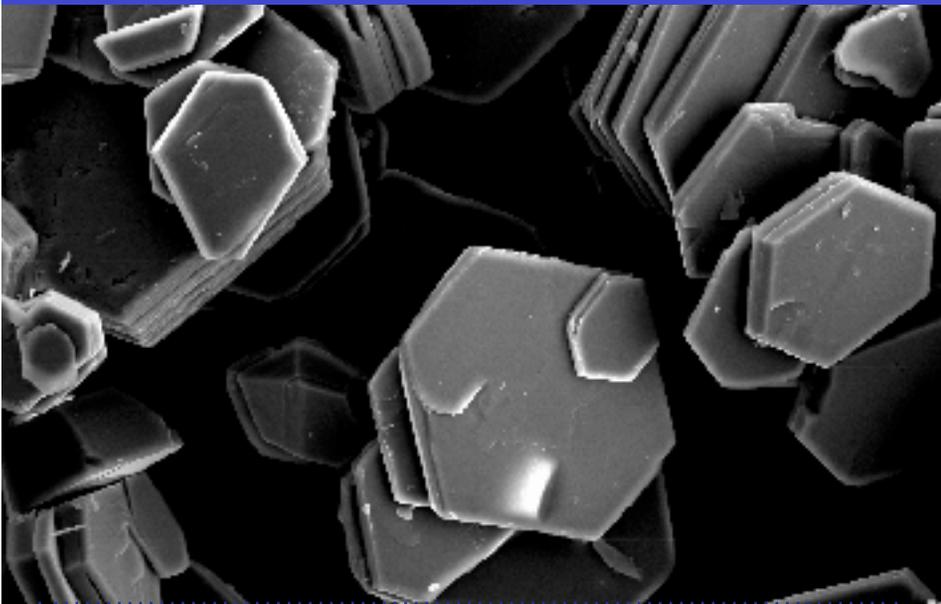
<http://www.gla.ac.uk/projects/originoflife/>)

Transition from prebiotic chemistry to early life: did mineral-catalyzed processes transform into metabolism??



(image courtesy Mike Russell;
<http://www.gla.ac.uk/projects/originoflife/>)

Minerals may have also served as templates for formation of polymers (peptides, nucleic acids) on the prebiotic Earth

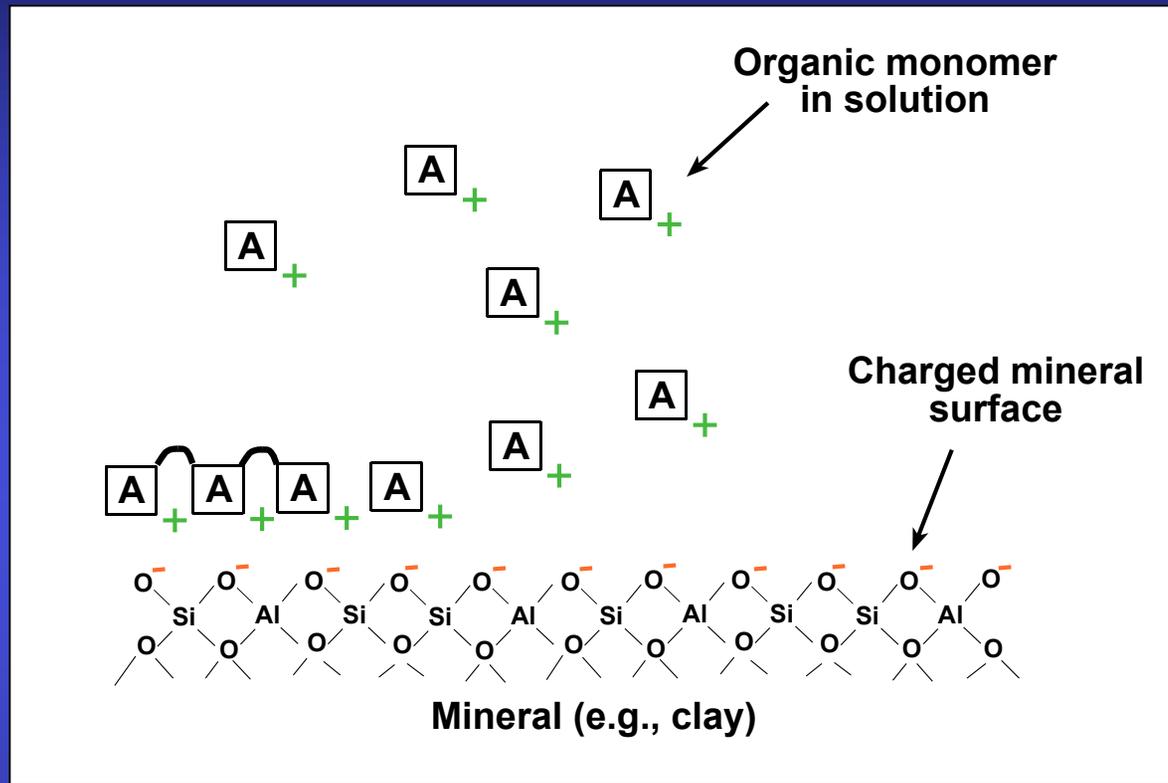


clay (microscopic image)

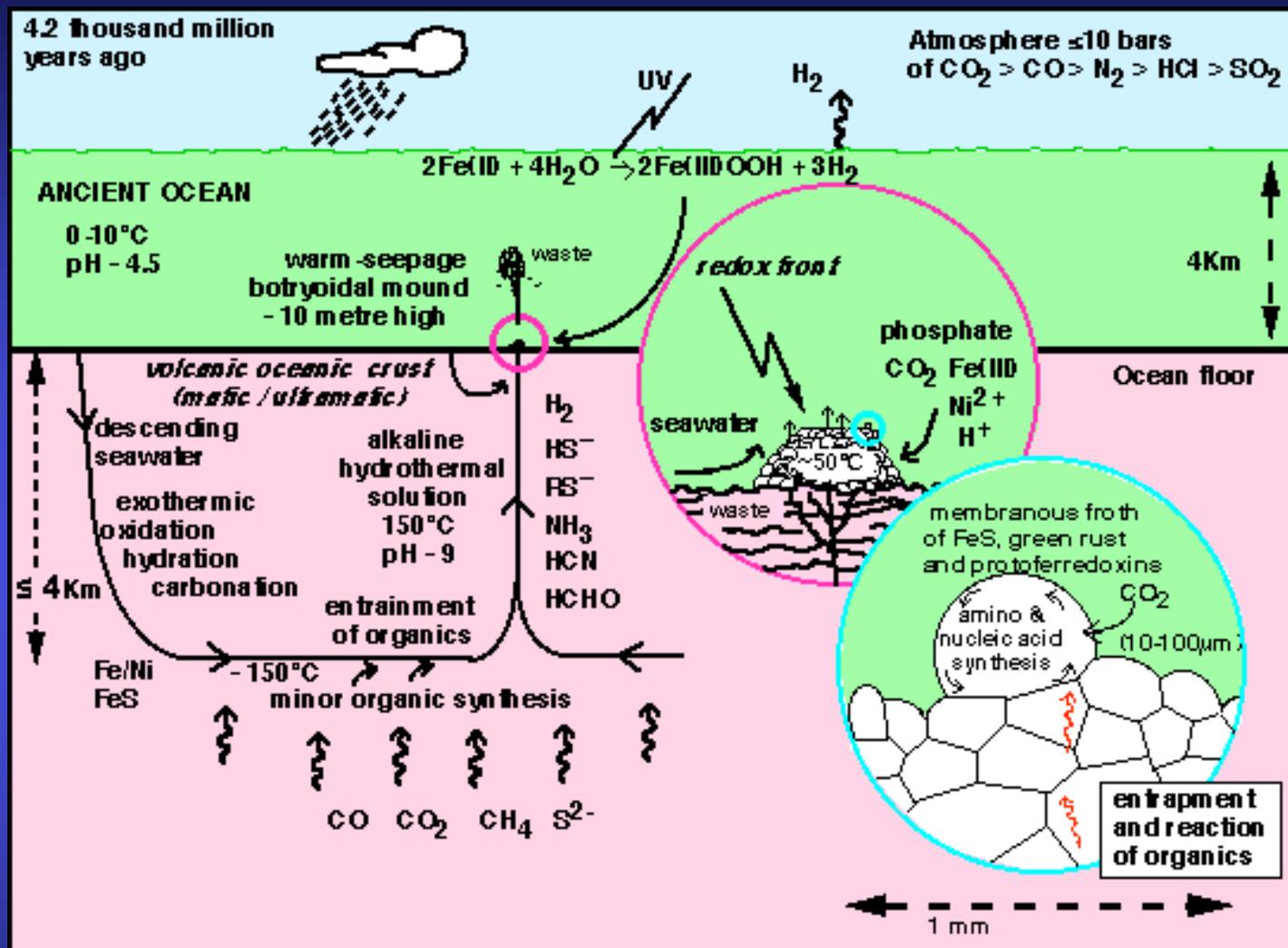


Pyrite (iron sulfide)

Minerals surfaces as templates:



Monomers in solution (amino acids, nucleotides) adhere to charged surface and polymerize, possibly forming precursors to proteins and RNA



(image courtesy Mike Russell;
<http://www.gla.ac.uk/projects/originoflife/>)

The early Earth

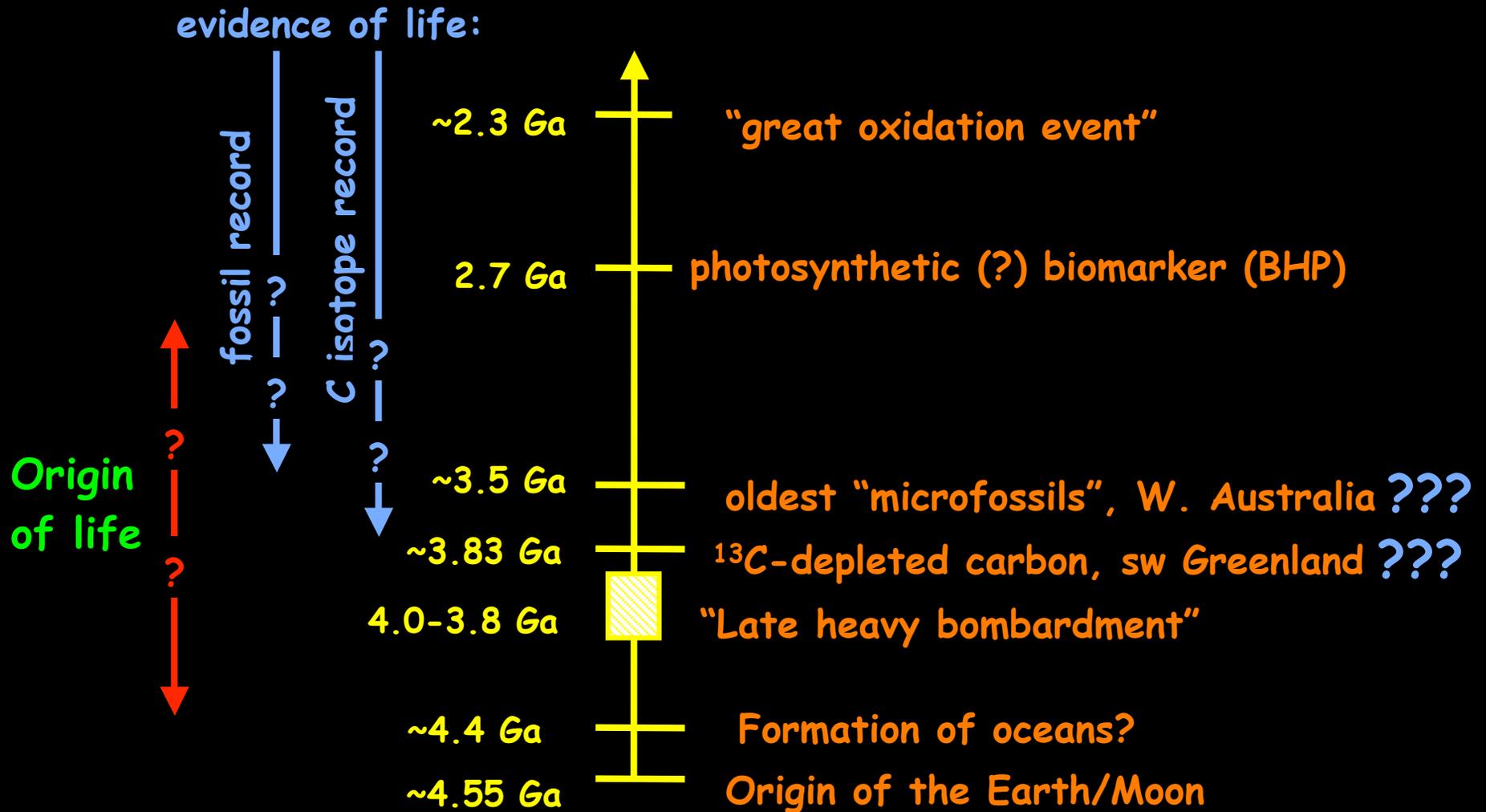


Smithsonian Institution

Oldest rocks on Earth <4 billion years old

⇒⇒ *(almost) no geologic record of the very Earth!!*

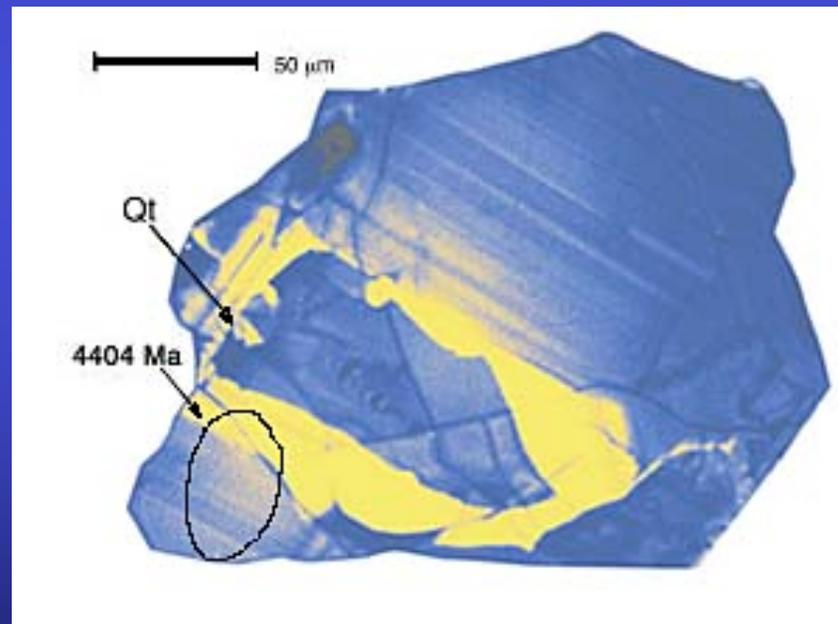
Timeline of the early Earth



Oldest rocks on Earth <4 billion years old.
But... older zircon minerals in younger rocks
provide some information on the first 500
million years.

Occurrence and
isotopic
composition of
oldest zircons
indicates
continents and
oceans were
present within a

~150 million years (from website of Dr. John Valley, Univ. of Wisconsin)
after Earth



What can we infer about hydrothermal environments on the very early Earth?

- ◆ Heat flow much greater, therefore hydrothermal environments were likely much more abundant (both submarine and continental)
- ◆ Atmosphere/oceans much less oxidizing
 - ⇒ electron acceptors for metabolic energy sources much *less* abundant
- ◆ Chemically reducing volcanic rocks more prevalent
 - ⇒ electron donors for metabolic energy sources much *more* abundant

Peridotite-hosted hydrothermal systems were more prevalent on the early Earth

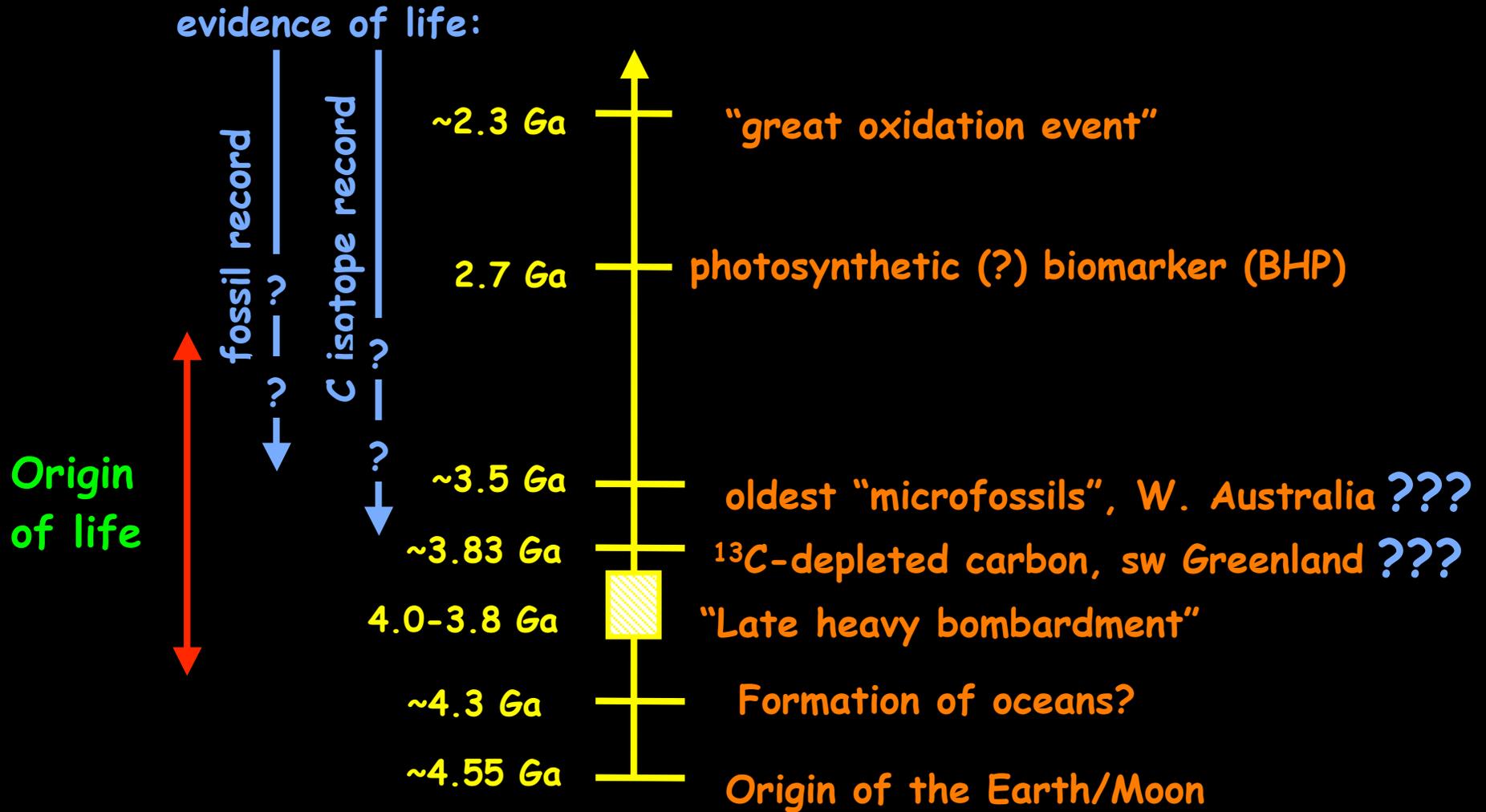


Produce alkaline fluids highly enriched in H_2 and methane.

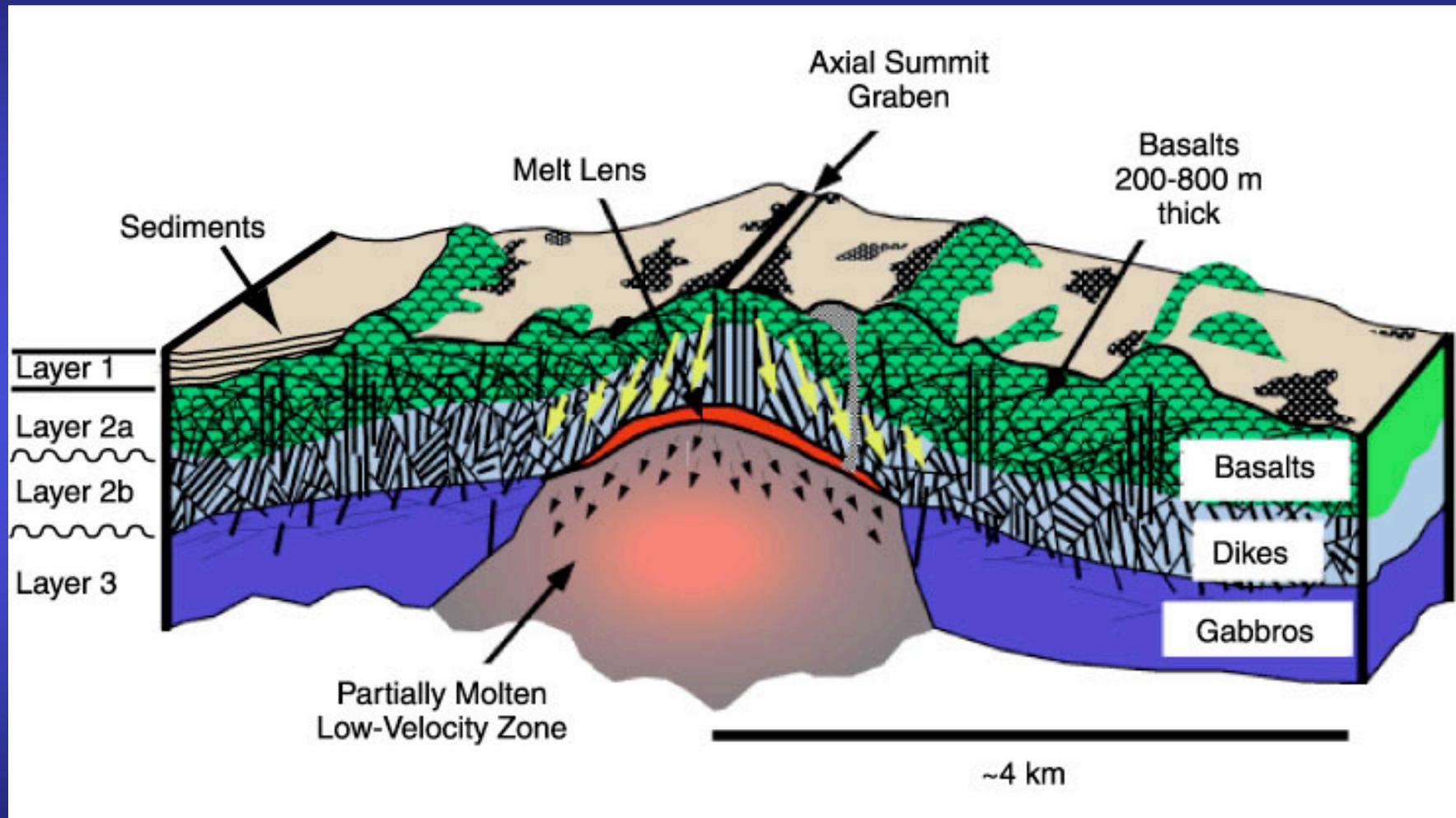


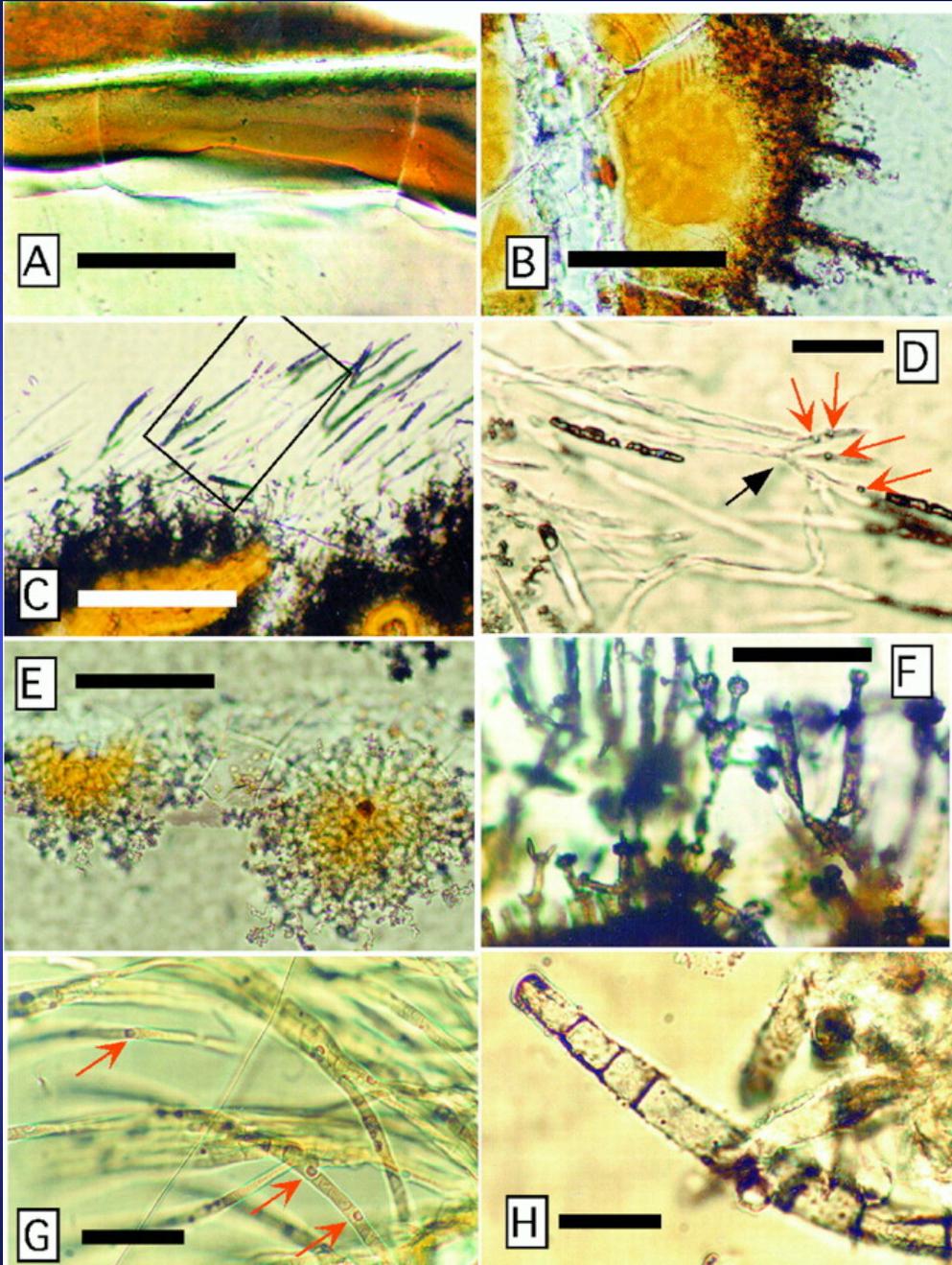
Chimney structures from Lost City, a modern peridotite-hosted system (Kelley et al., *Science*, 2005)

Timeline of the early Earth



Cross-section of the ocean crust at mid-ocean ridge

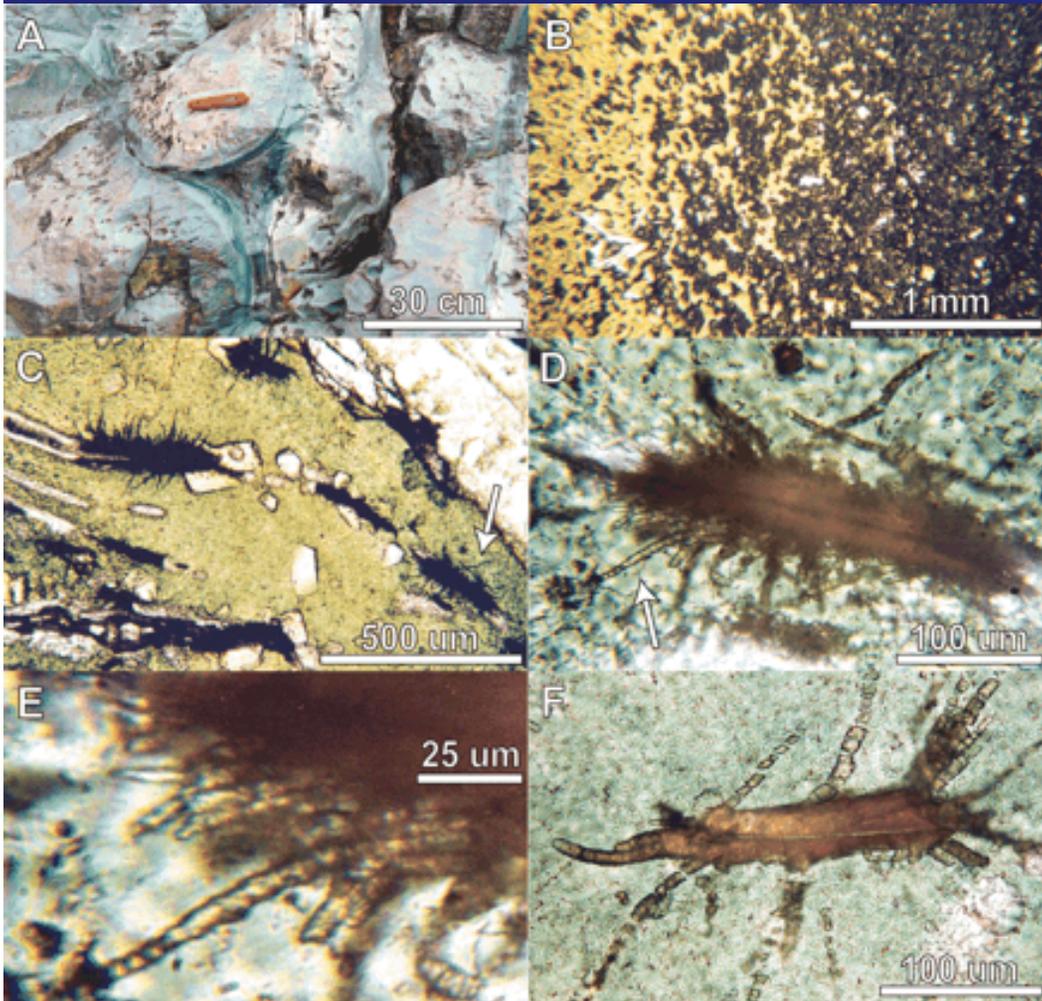




Subsurface life in the ocean crust

Morphological & chemical
evidence for active
microorganisms 100's of
meters beneath seafloor

(Fisk et al., *Science*, 1998)



Microbial(?) textures in
~3.5 billion year old
rocks,
Barberton Greenstone Belt

(Furnes et al., *Science*, 2004)

Chemolithoautotrophic iron-oxidizing bacteria growing on the mineral pyrite as an energy source (also grows very well on basalt)

